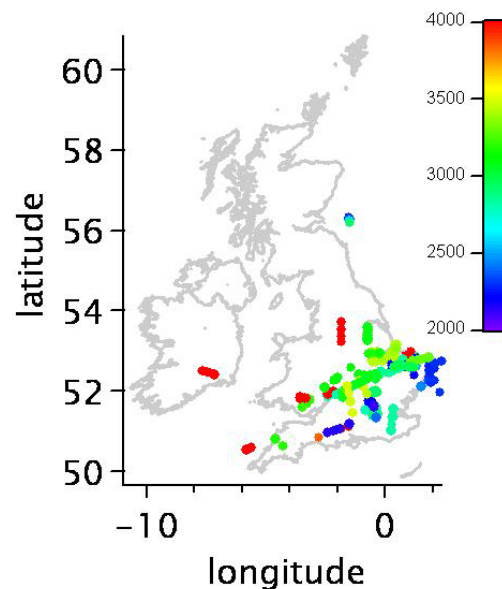


Long-term monitoring of VOCs in the free troposphere above the UK

Shalini Punjabi^a, James R. Hopkins^{a,b}, Alastair Lewis^{a,b}

^a University of York, York ^bNCAS, University of York, York

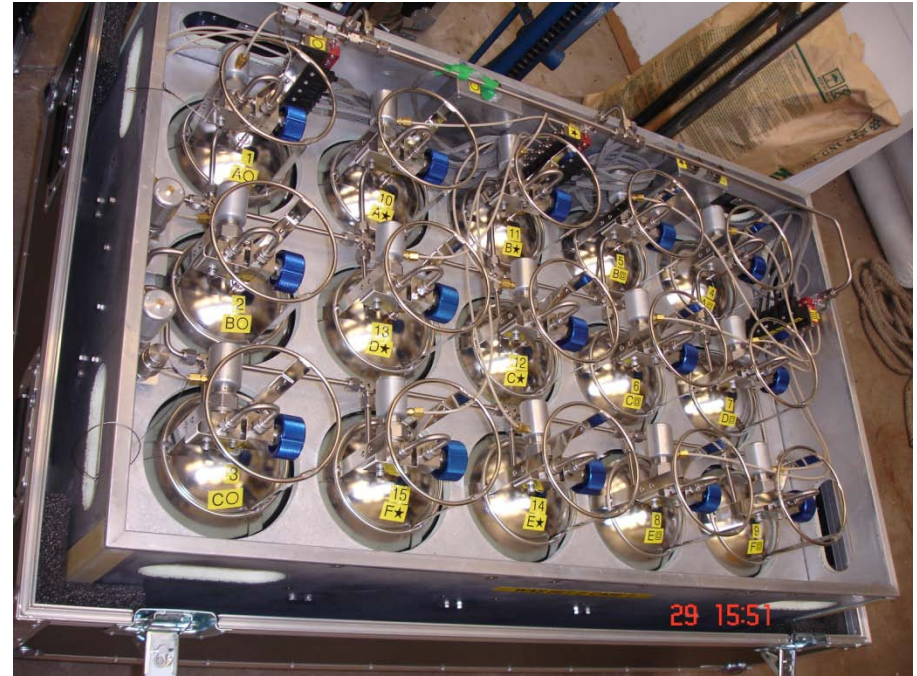


VOC monitoring : Aims

- To characterize seasonal and inter annual trends
- To capture important features related to transport, mixing and chemistry.
- To utilise UK based research aircraft's flying hours to the fullest for long-term study.

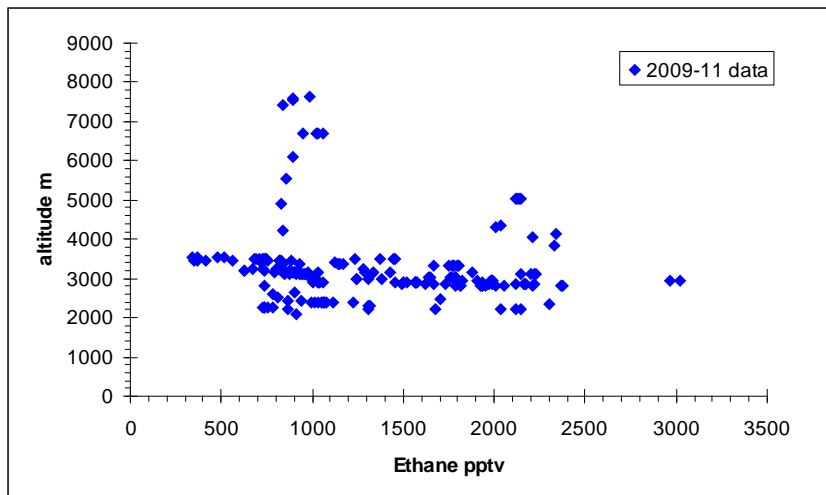
Sample collection

- Silico-treated stainless steel canisters
- Max. filling pressure 40psi
- Altitude preferably : 2000 - 5000m



The Whole Air Sampling System

- Sampling frequency : variable, typically once in a month

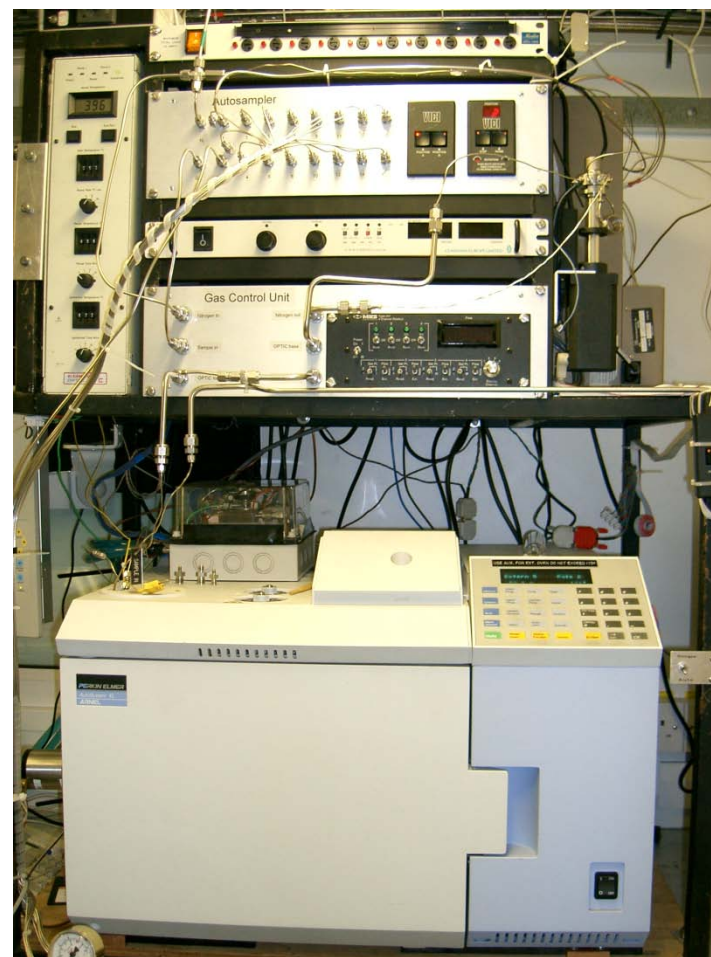


Technique and analytical configuration

York FGAM dc-gc-fid

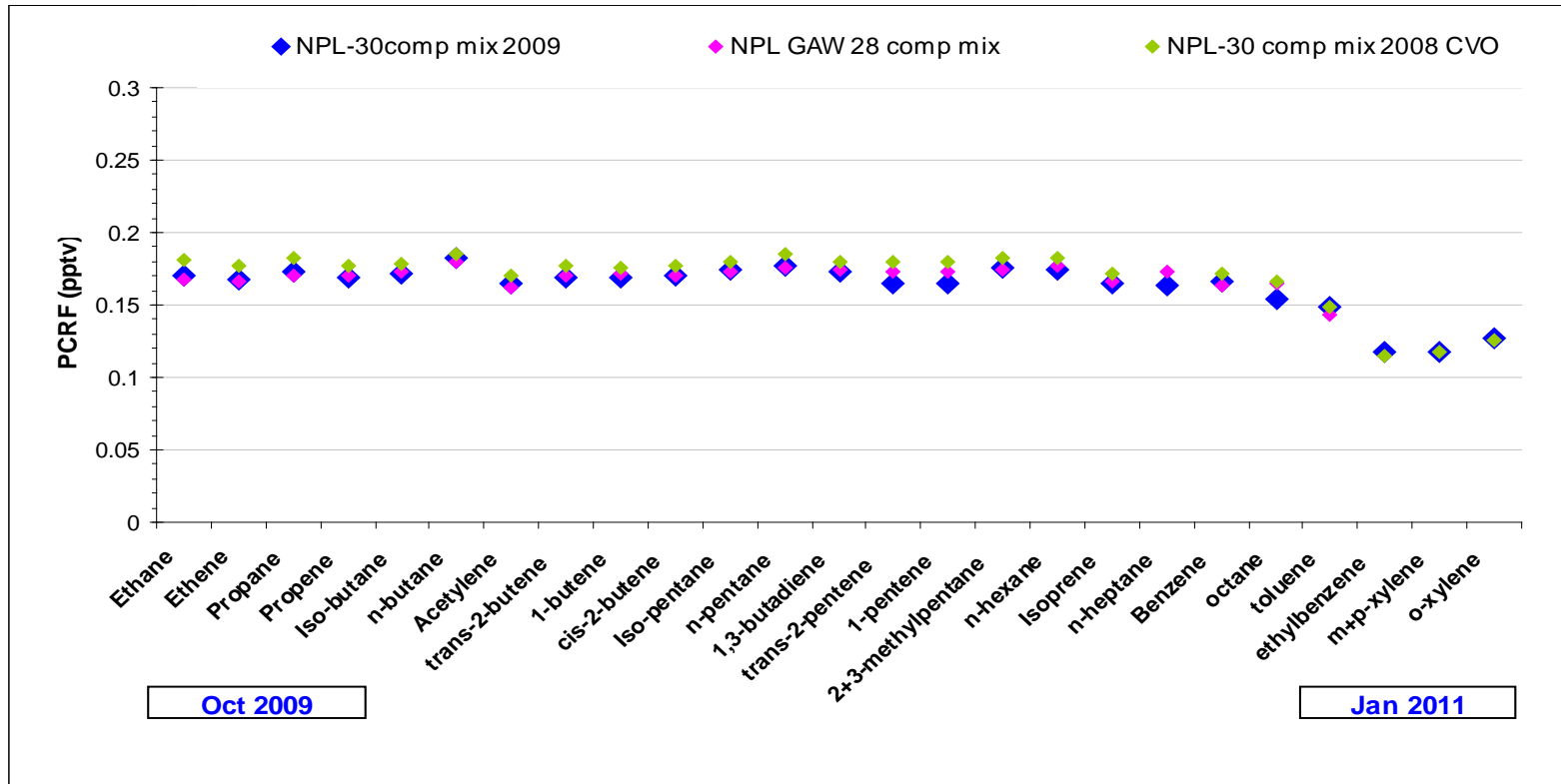
Dual channel gas chromatograph

- Water removal
 - condensation trap maintained at -30°C
- Preconcentration
 - Trapping temperature : Peltiers cooling -23°C
 - Multi adsorbent bed: Carboxen B and carboxen 1000
- Thermal desorption
 - 320°C with carrier flow (helium)
- Separation
 - Two capillary columns 50 m PLOT and 10 m LOWOX
- Detection
 - FID 1 and FID 2
- Calibration and standards:
 - National Physical Laboratory (NPL)-30 component mix
 - GAW-VOC standards



C₂-C₈ Alkanes, C₂-C₅ Alkenes
Aromatics (BTEX), Acetylene
C₂-C₄ Oxygenated VOCs

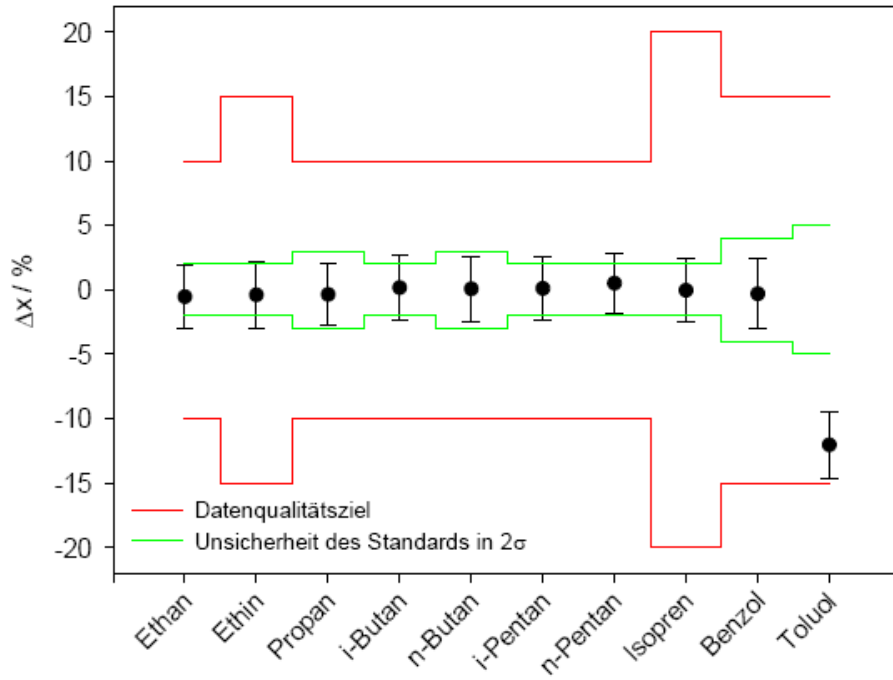
System performance with time



WCC-VOC intercomparison

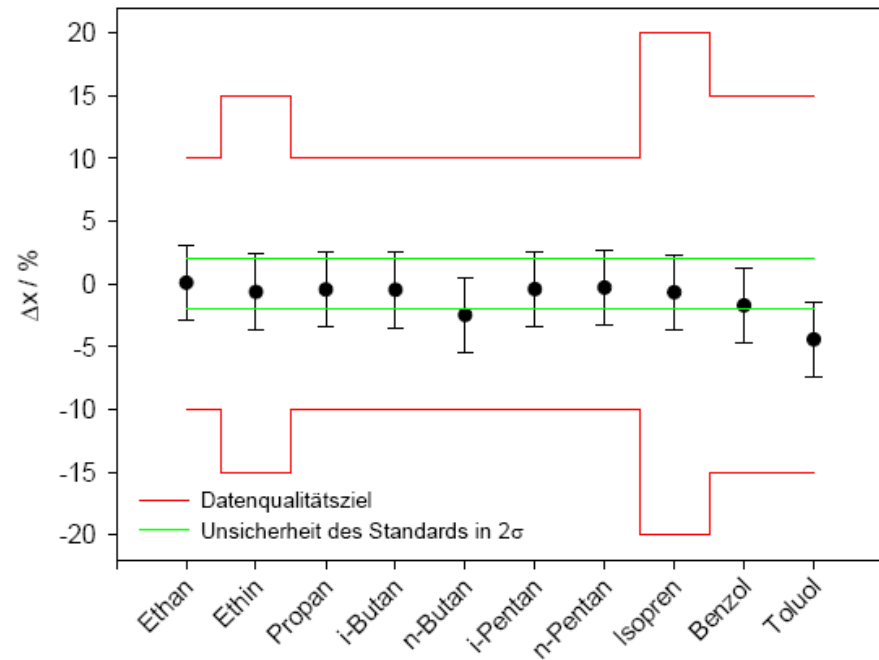
York D336442

2010

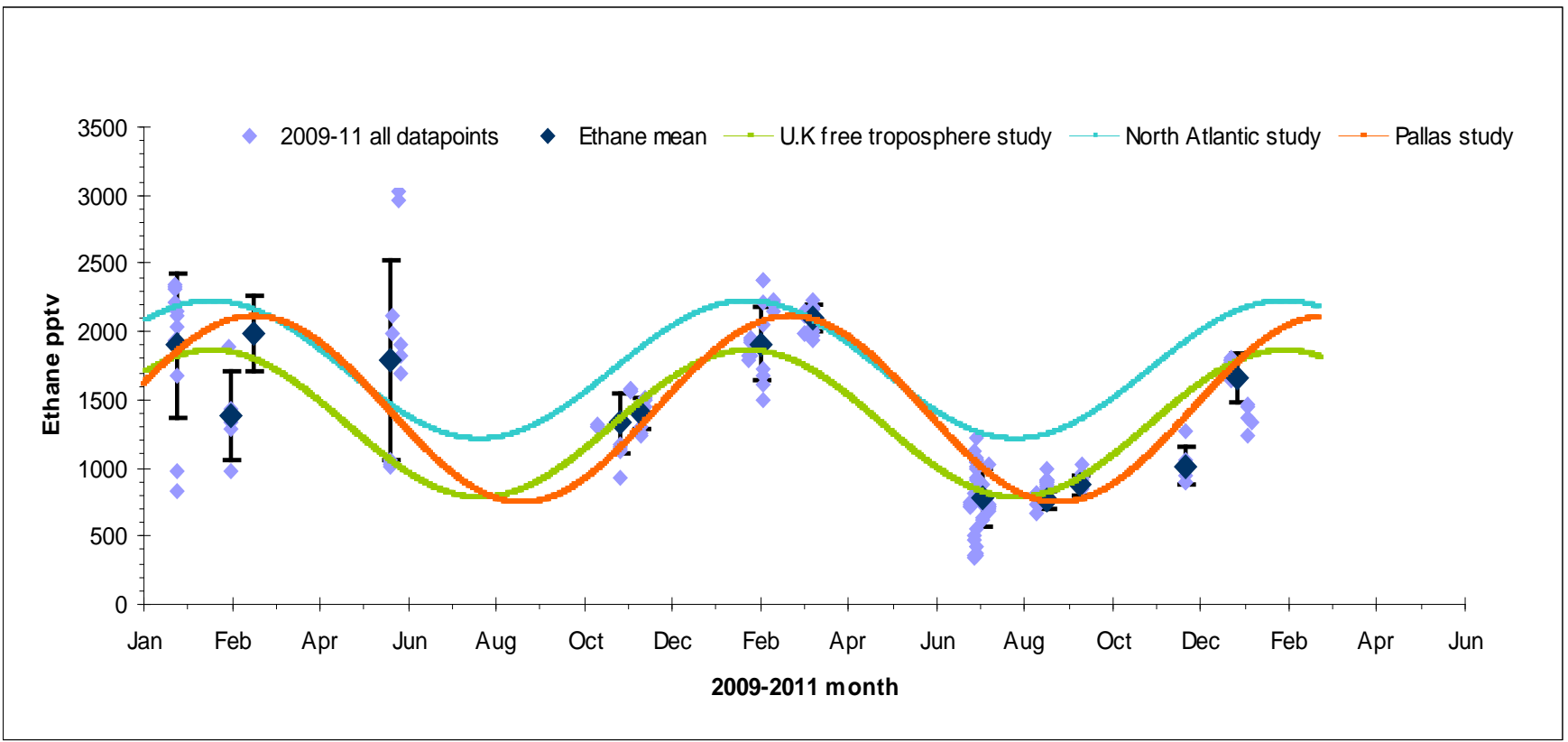


York D292363

2010



Seasonal trends : Ethane



U.K free troposphere,2009-11
North Atlantic study,1989
Pallas ,Baltic sea ,1994

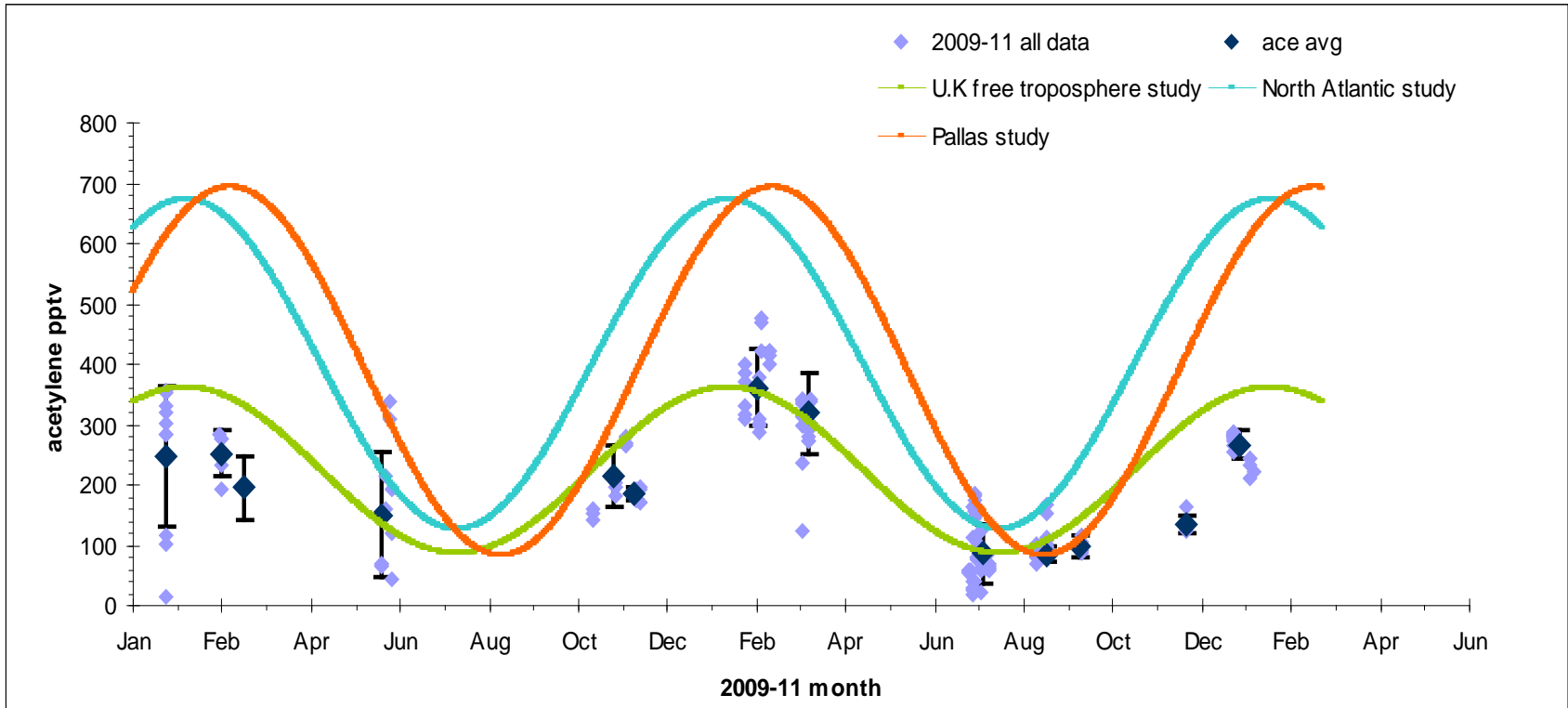
$$y=1318.32 + 539.34(2\pi/365(\text{day}+46.25))$$

$$y = 1715 + 505(2\pi/365(\text{day}+46.25))$$

$$y=1425.00 + 683.34(2\pi/365(\text{day}+16.25))$$

Laurila, T. and Hakola, H., 1996 ,Atmospheric Environment 30, pp. 1597–1607
 S.A. Penkett, N.J. Blake, P. Lightman, A.R.W. Marsh and P. Anwyl, Journal of Geophysical Research 98 (1993), pp. 2865–2885

Seasonal trends: Acetylene



U.K free troposphere study ,2009-11

North Atlantic study ,1989

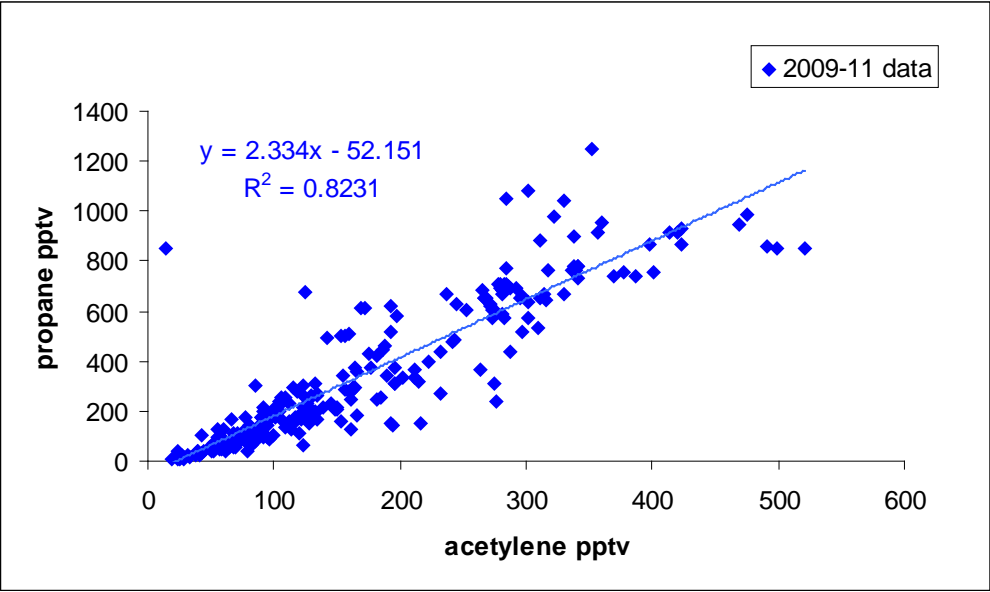
Pallas study ,1994

$$y = 223.67 + 137.47 (2\pi/365(\text{day}+56.25))$$

$$y = 399.5 + 273.5 (2\pi/365(\text{day}+56.25))$$

$$y = 388 + 305.33.5 (2\pi/365(\text{day}+26.25))$$

Ratios and source profile variation



$$\ln[P]/\{A\} = (k_p - k_a/k_p) \ln[P] + C$$

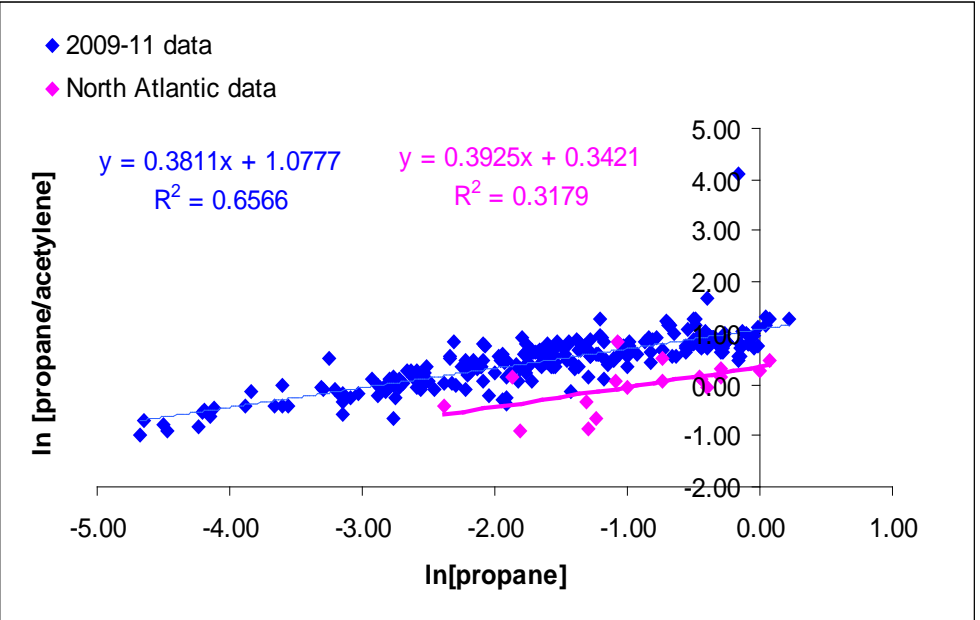
↑
slope=rate constants

Where $C = \ln [P]_0 / [A]_0 - (k_p - k_a/k_p) \ln[P]_0$

↑
intercept= source profile

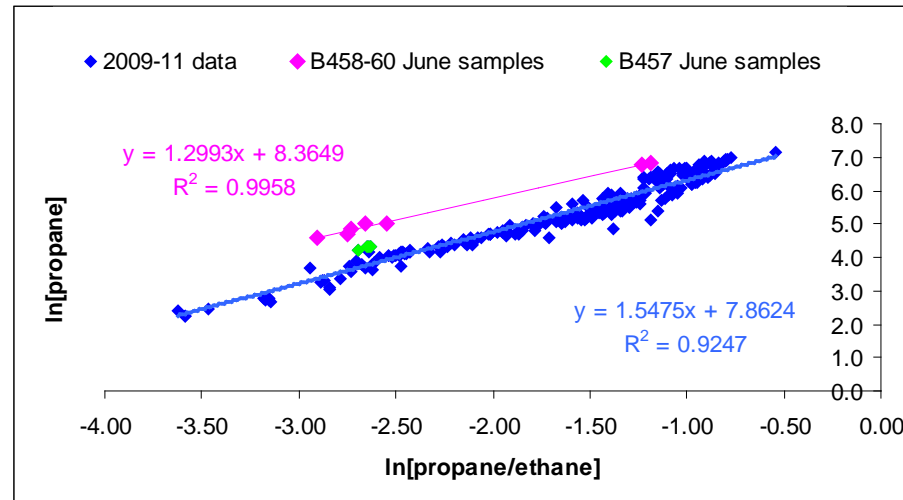
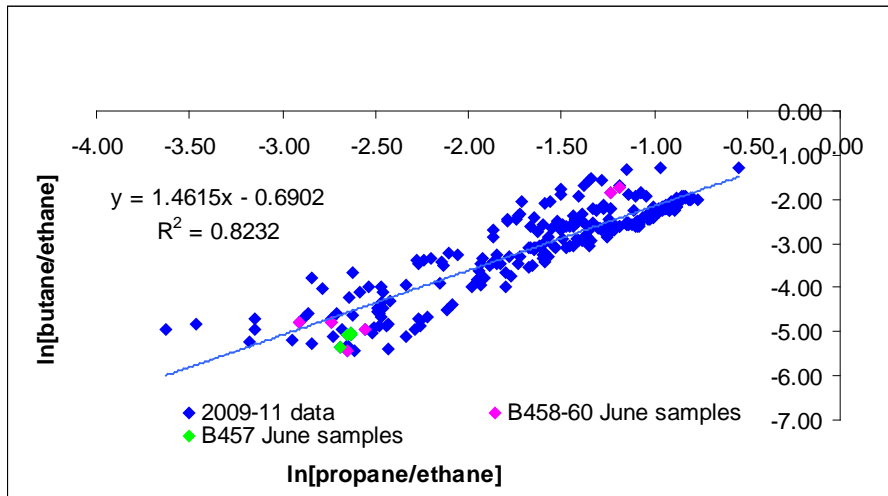
Our study and intercept = 1.07

North Atlantic study and intercept = 0.34



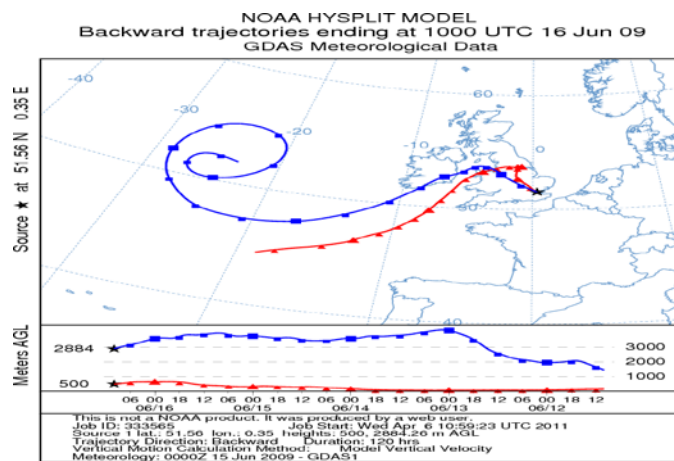
Goldan, P. D., D. D. Parrish, W. C. Kuster, M. Trainer, S. A. McKeen, J. Holloway, B. T. Jobson, D. T. Sueper, and F. C. Fehsenfeld (2000), *J. Geophys. Res.*, 105(D7), 9091–9105, doi:10.1029/1999JD900429.
Gong, Q., and K. Demerjian (1997), *J. Geophys. Res.*, 102(D23), 28059-28069.

Ratios and aging pattern: another source

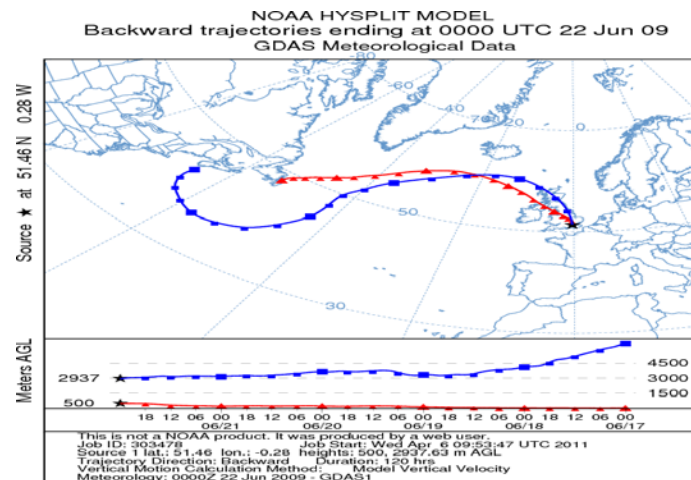


In [butane/ethane] vs In[propane/ethane]

In [propane/ethane] vs In [propane]

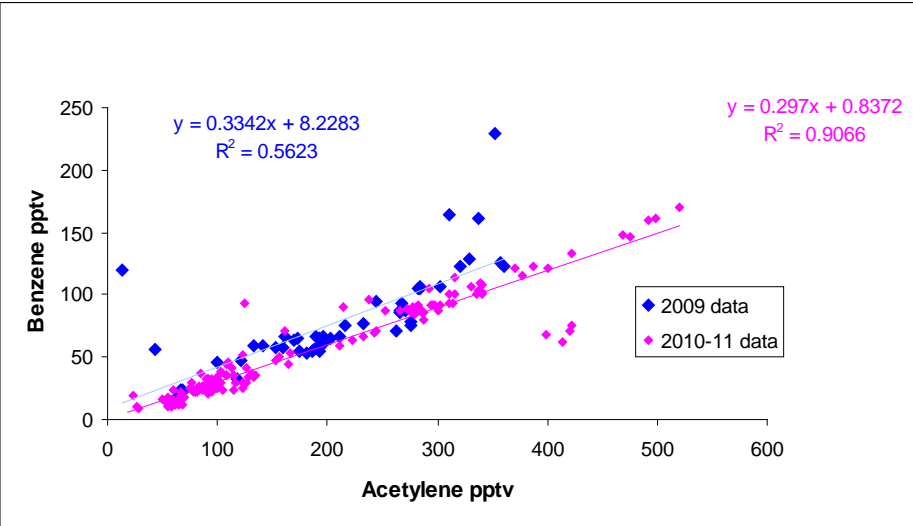


Flight B457 Atlantic ocean



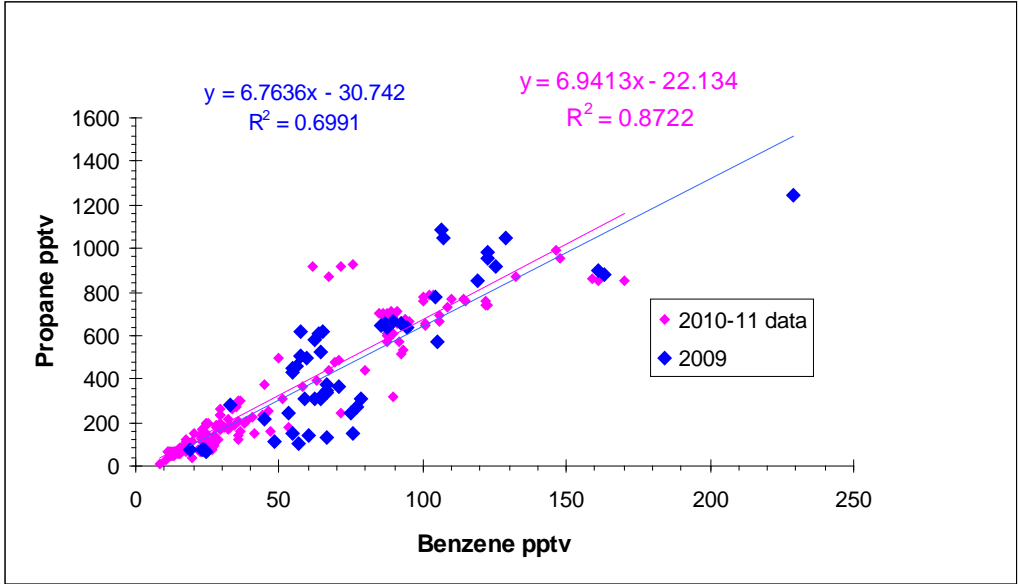
Flight B458 continental influence

Ratios : source characterization

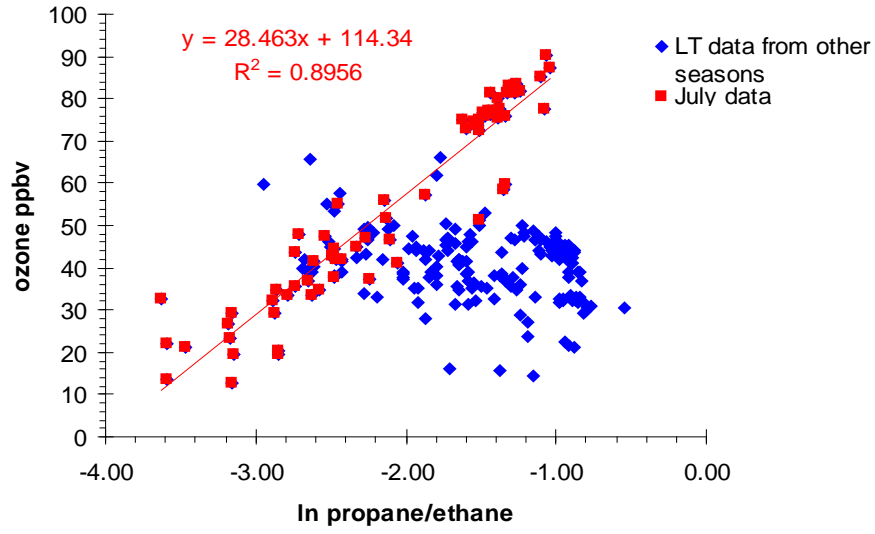


Vehicular exhaust responsible for benzene and acetylene emissions.

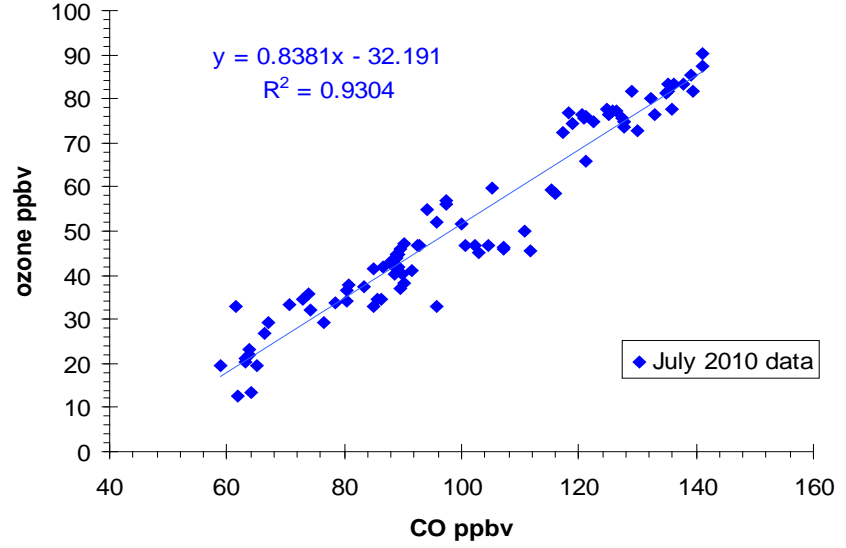
Natural gas as a main source for propane emission



Hydrocarbon Processing and ozone



Low NO levels or mixing of polluted air with clean marine air could be the reasons for ozone destruction.



Helmig, D., D. M. Tanner, R. E. Honrath, R. C. Owen, and D. D. Parrish (2008), *J. Geophys. Res.*, 113, D20S91, doi:10.1029/2007JD008930.

Summary and further plans

- Hydrocarbon ratios as a tracer
- Free tropospheric study provides useful information on transport, mixing and chemistry
- Continue to collect data over longer timescales and monitor instrument performance
- Improve data coverage by collecting and analysing more samples.

Acknowledgements

My supervisors Ally Lewis and Jim Hopkins

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Research group people

