

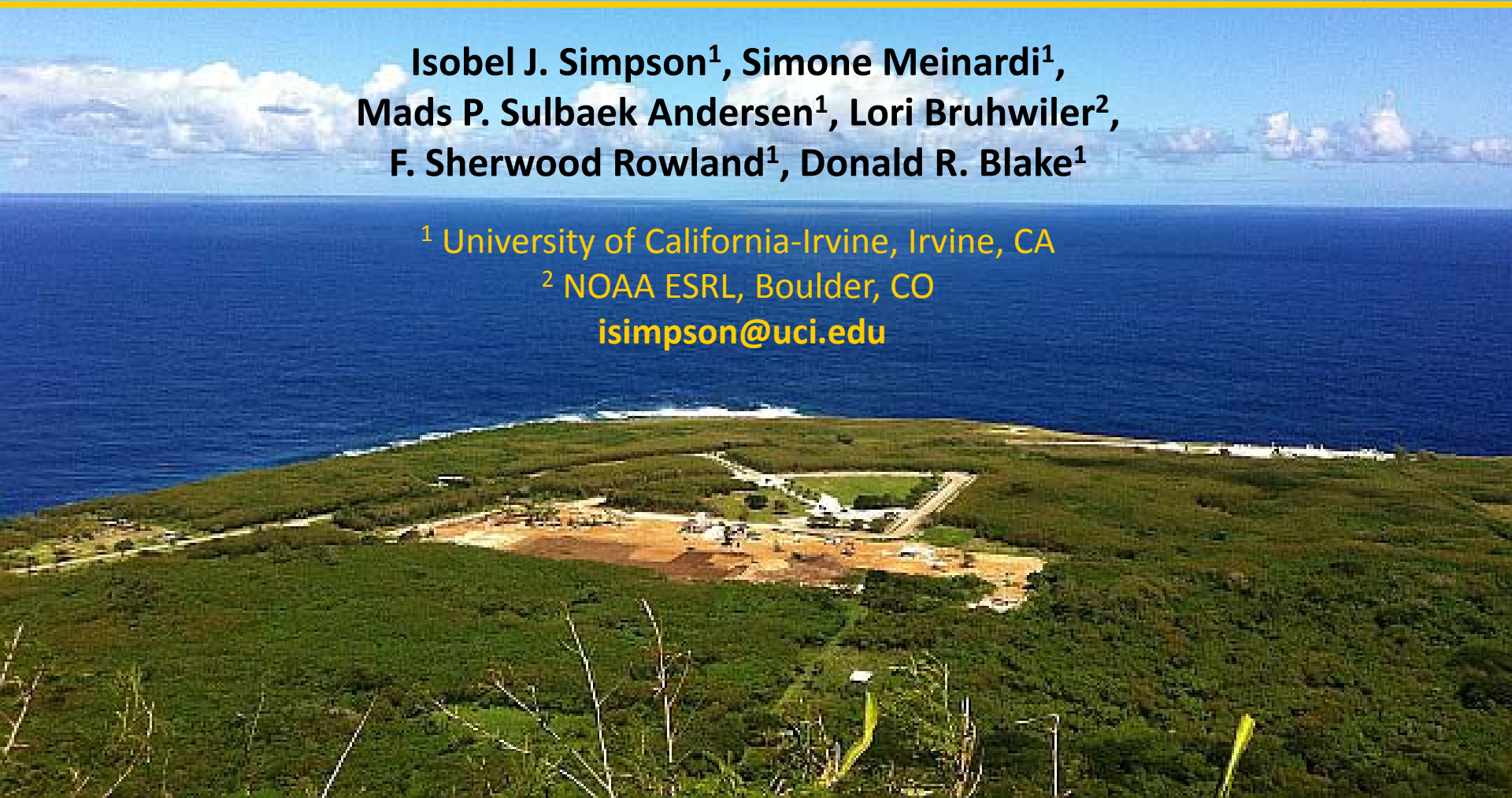
# Highlights of the UC-Irvine Global Trace Gas Monitoring Program (1978-2010)

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# UC-Irvine whole air sampling (WAS)



*Canister sampling in Rarotonga (21°S)*

## Canister-based global monitoring

- 2-L stainless steel
- Conditioned, evacuated
- Bellows valve
- Sampling period: 1 minute
- Sampling pressure: ambient



*Nugget Point, New Zealand (46°S)*

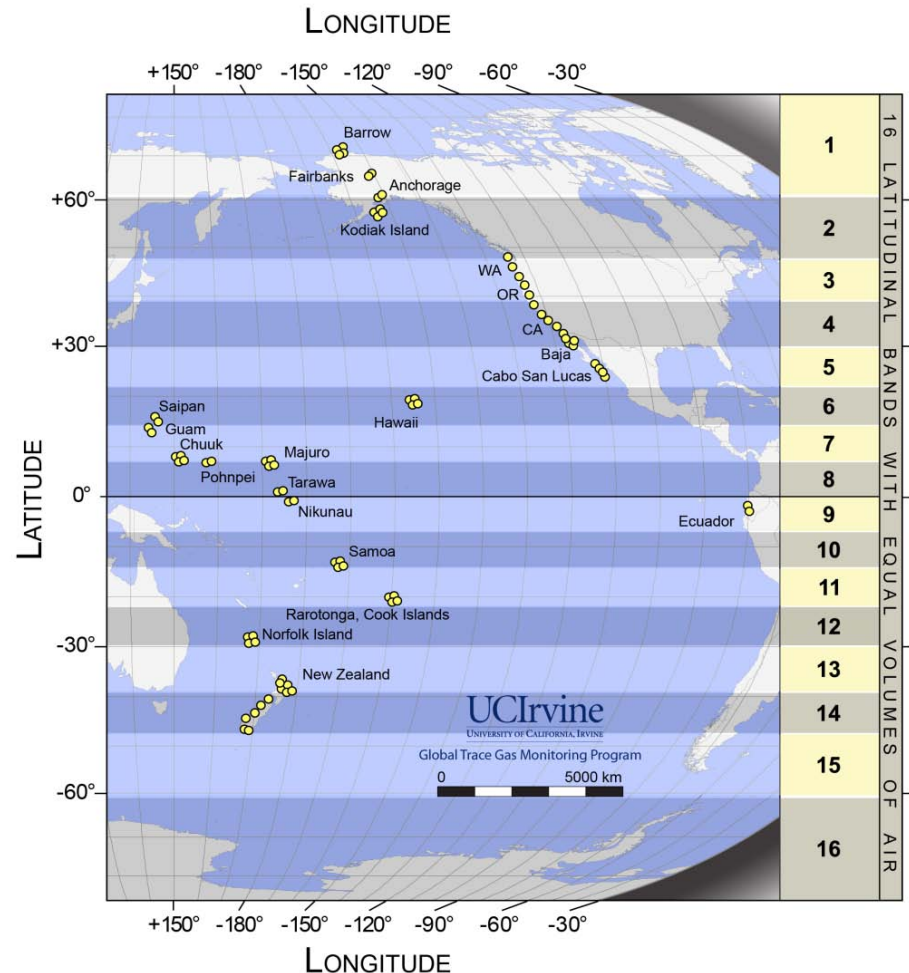
# Seasonal sampling in the Pacific Basin

## Seasonal sampling

- 80 samples per season
- 40-45 locations
- 3-week period

## Global averaging

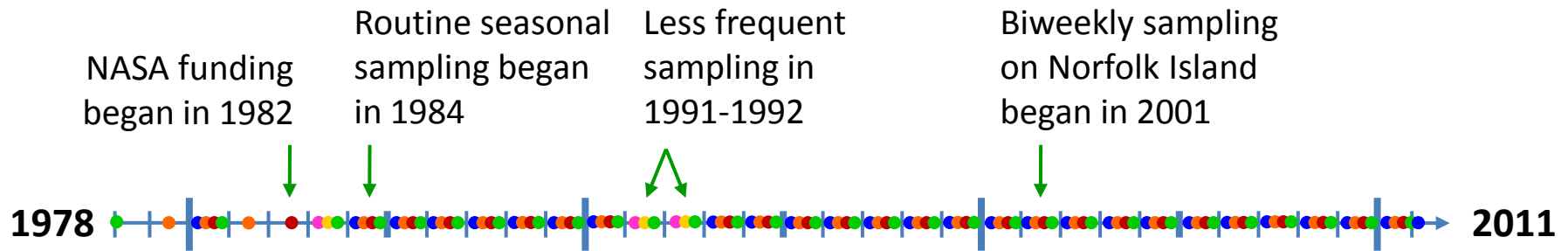
- 16 latitudinal bands
  - Each with equal volume of air
- Global averaging:
  1. Average within each band
  2. Interpolate box 15 & 16 values
  3. Average 16 band averages



J F M A M J J A S O N D



# Timeline of global trace gases monitored by UC-Irvine



**1978:**

- CH<sub>4</sub>
- CFC-11
- CFC-12
- CFC-113
- CH<sub>3</sub>CCl<sub>3</sub>
- CCl<sub>4</sub>



**1984:**

Ethane

**1988:**

C<sub>2</sub>Cl<sub>4</sub>

● ● ●
● March
● June
● September
● December
● ●
● April
● August
● December

**1996:**

- CHCl<sub>3</sub>
- H-1211
- Propane
- i*-Butane
- n*-Butane
- Ethyne
- EtONO<sub>2</sub>
- i*-PrONO<sub>2</sub>

**2001:**

MeONO<sub>2</sub>



Norfolk Island (29°S)

# Laboratory analysis at UC-Irvine

## Gas Chromatography (GC)

### Flame Ionization Detection (FID)

- Sensitive to hydrocarbons

### Electron Capture Detection (ECD)

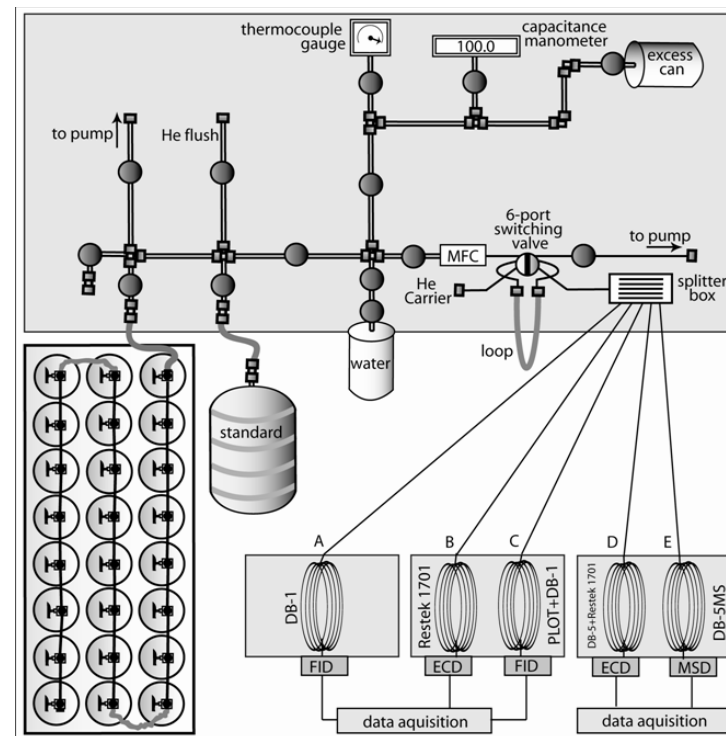
- Sensitive to halocarbons,  $\text{RONO}_2$

### Mass Spectrometer Detection (MSD)

- Unambiguous compound identification



Laboratory analysis performed by Brent Love



<u>Compound</u>	<u>LOD</u>	<u>Precision</u>	<u>Accuracy</u>
Methane		1 ppbv	1%
Ethane	3 pptv	1%	5%
CFC-11	10 pptv	1%	5%
$\text{CHCl}_3$	0.1 pptv	5%	5%
$\text{C}_2\text{Cl}_4$	0.01 pptv	2%	5%
$\text{MeONO}_2$	0.01 pptv	3%	10%

# UC-Irvine C<sub>1</sub>-C<sub>10</sub> VOC measurements

## Hydrocarbons

1. Methane
2. Ethane
3. Ethene
4. Ethyne
5. Propane
6. Propene
7. Propyne
8. *i*-Butane
9. *n*-Butane
10. 1-Butene
11. *i*-Butene
12. *t*-2-Butene
13. *c*-2-Butene
14. 1,3-Butadiene
15. *i*-Pentane
16. *n*-Pentane
17. Isoprene
18. *n*-Hexane
19. *n*-Heptane
20. *n*-Octane
21. *n*-Nonane

## Hydrocarbons

22. 2,3-Dimethylbutane
23. 2+3-Methylpentane
24. Cyclopentane
25. Methylcyclopentane
26. Cyclohexane
27. Methylcyclohexane
28. Benzene
29. Toluene
30. Ethylbenzene
31. *m+p*-Xylene
32. *o*-Xylene
33. Styrene
34. *n*-Propylbenzene
35. 2-Ethyltoluene
36. 3-Ethyltoluene
37. 4-Ethyltoluene
38. 1,3,5-Trimethylbenzene
39. 1,2,4-Trimethylbenzene
40. 1,2,3-Trimethylbenzene
41.  $\alpha$ -Pinene
42.  $\beta$ -Pinene

## Hydrocarbons

43. Furan
44. Methanol
45. Ethanol
46. Acetone
47. Acetaldehyde
48. MEK
49. MAC
50. MVK
51. MTBE

## Alkyl Nitrates

52. MeONO<sub>2</sub>
53. EtONO<sub>2</sub>
54. *i*-PrONO<sub>2</sub>
55. *n*-PrONO<sub>2</sub>
56. 2-BuONO<sub>2</sub>
57. 2-PeONO<sub>2</sub>
58. 3-PeONO<sub>2</sub>
59. 3-Me-2-BuONO<sub>2</sub>

## Sulfur Species

60. OCS
61. DMS

## Halocarbons

62. CFC-11
63. CFC-12
64. CFC-113
65. CFC-114
66. H-1211
67. H-1301
68. H-2402
69. HFC-134a
70. HFC-152a
71. HCFC-22
72. HCFC-141b
73. HCFC-142b
74. CCl<sub>4</sub>
75. CH<sub>3</sub>CCl<sub>3</sub>
76. CH<sub>2</sub>Cl<sub>2</sub>
77. C<sub>2</sub>HCl<sub>3</sub>
78. CHCl<sub>3</sub>
79. C<sub>2</sub>Cl<sub>4</sub>

80. CH<sub>3</sub>Cl
81. CH<sub>3</sub>Br
82. CH<sub>3</sub>I
83. CHBr<sub>2</sub>Cl
84. CHBrCl<sub>2</sub>

## Halocarbons

85. CH<sub>2</sub>Br<sub>2</sub>
86. CHBr<sub>3</sub>
87. Ethylchloride
88. 1,2-DCE

**Alkanes**  
**Alkenes**  
**Alkynes**  
**Cycloalkanes**  
**Aromatics**  
**Monoterpenes**  
**Oxygenates**  
**Alkyl nitrates**  
**Sulfur species**  
**Halocarbons**



# Latitudinal and seasonal trends

## Latitudinal trends

- **Anthropogenic sources:**
  - North/South gradient
  - $\text{CH}_4$ ,  $\text{C}_2\text{Cl}_4$ , CFCs, **ethane** ...
- **Oceanic sources:**
  - Tropical maximum
  - **MeONO<sub>2</sub>**, EtONO<sub>2</sub> ...

## Seasonal trends

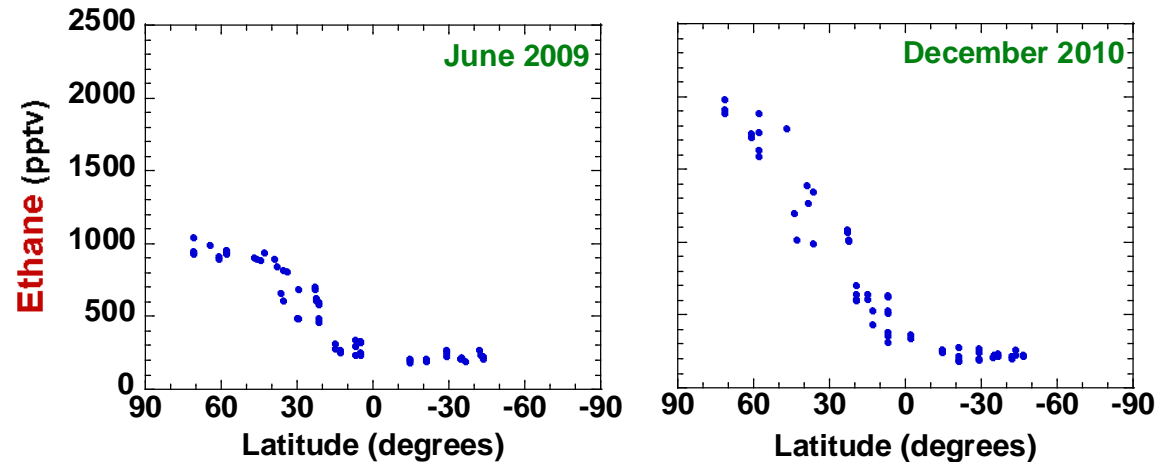
- Many species: major OH sink
- Winter maximum
- Summer minimum

### Atmospheric lifetimes:

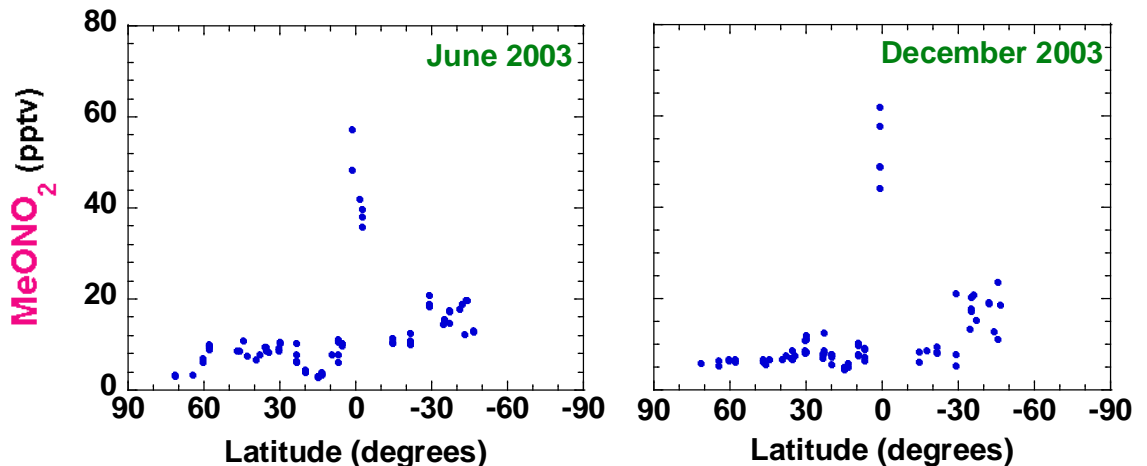
**Ethane:** 2-3 mo

**Methyl nitrate:** 1 mo

## Ethane: Strong fossil fuel/biomass burning sources



## MeONO<sub>2</sub>: Strong equatorial oceanic source



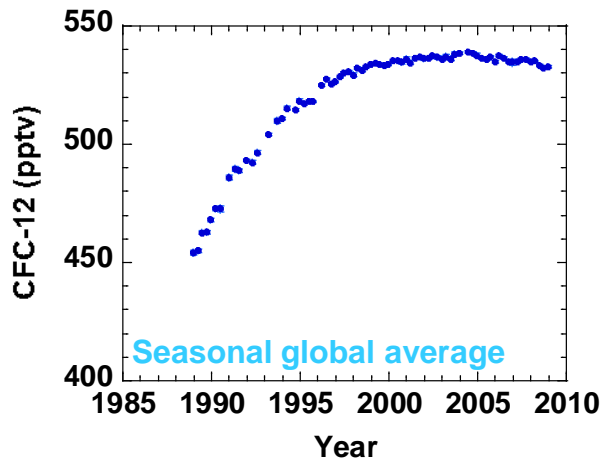
# Global trends of long-lived halocarbons

## CFC-12 ( $\text{CCl}_2\text{F}_2$ )

Lifetime: **100 yrs**

Peak: ~2004

Decline:  $-2.5 \text{ pptv yr}^{-1}$   
in 2008

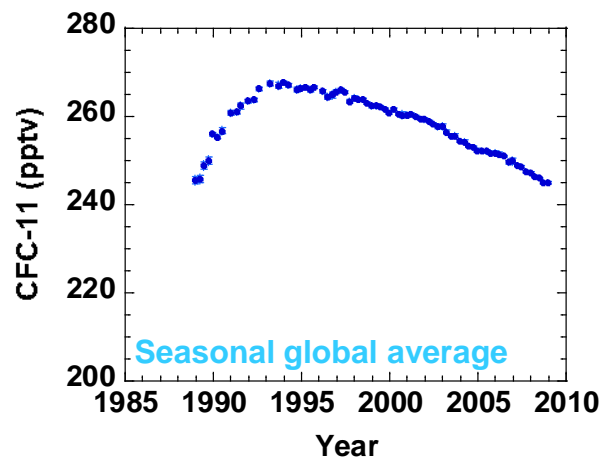


## CFC-11 ( $\text{CCl}_3\text{F}$ )

Lifetime: **45 yrs**

Peak: ~1993

Decline:  $-1.9 \text{ pptv yr}^{-1}$   
in 2008

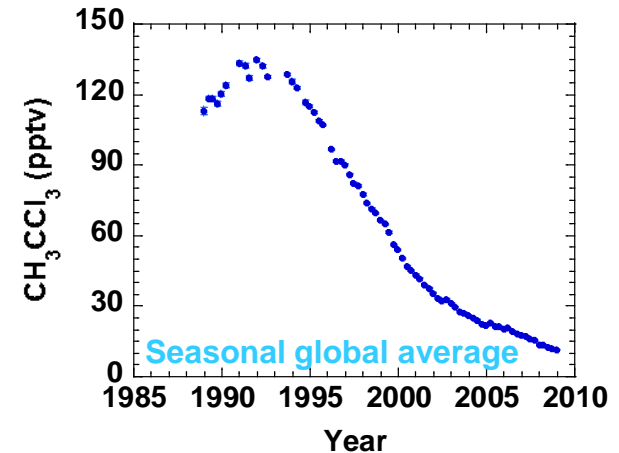


## $\text{CH}_3\text{CCl}_3$

Lifetime: **5 yrs**

Peak: ~1991

Decline:  $-2.2 \text{ pptv yr}^{-1}$   
in 2008



### Group CFC-12 (2008)\*

UCI	532.6 ppt
NOAA	535.5 ppt
AGAGE	537.4 ppt

### Group CFC-11 (2008)\*

UCI	244.2 ppt
NOAA	244.8 ppt
AGAGE	243.4 ppt

### Group $\text{CH}_3\text{CCl}_3$ (2008)\*

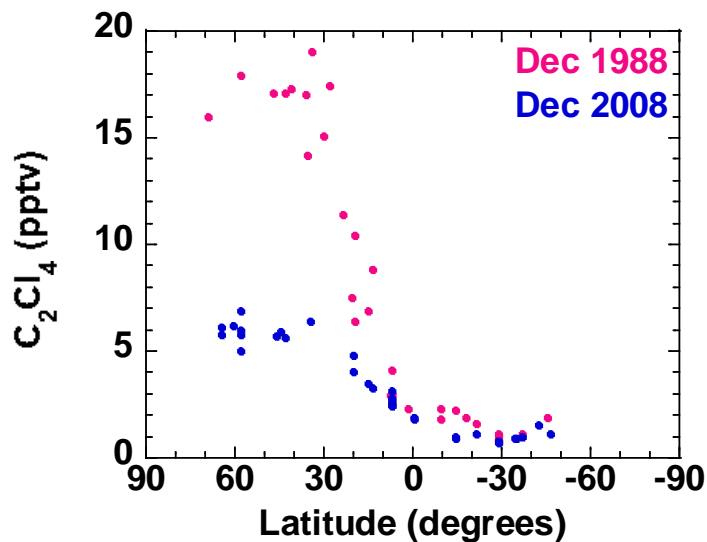
UCI	11.5 ppt
NOAA	11.1 ppt
AGAGE	10.7 ppt



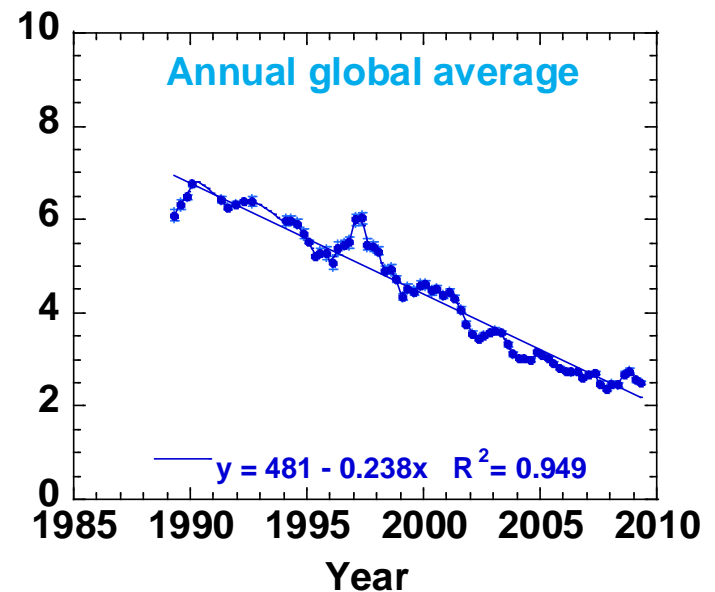
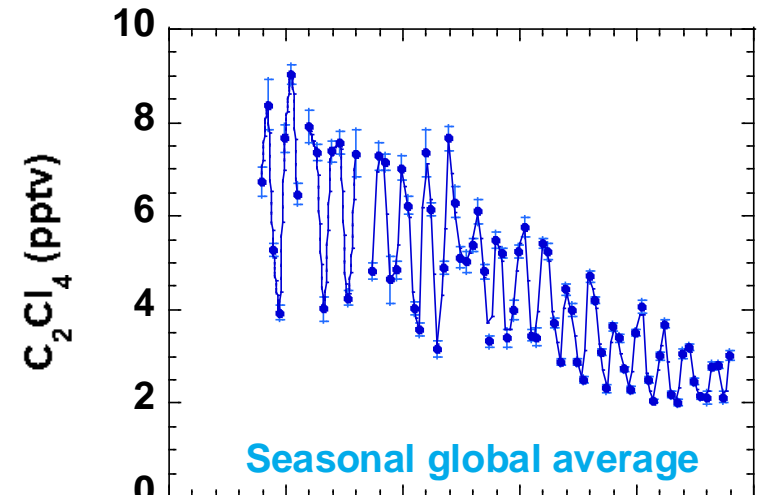
# Global $C_2Cl_4$ trend: Declining

## Long-term global $C_2Cl_4$ decline

- $6.3 \pm 0.1$  pptv in 1989
- $2.5 \pm 0.1$  pptv in 2009
- **3.8 pptv (60%) decline in 20 years**



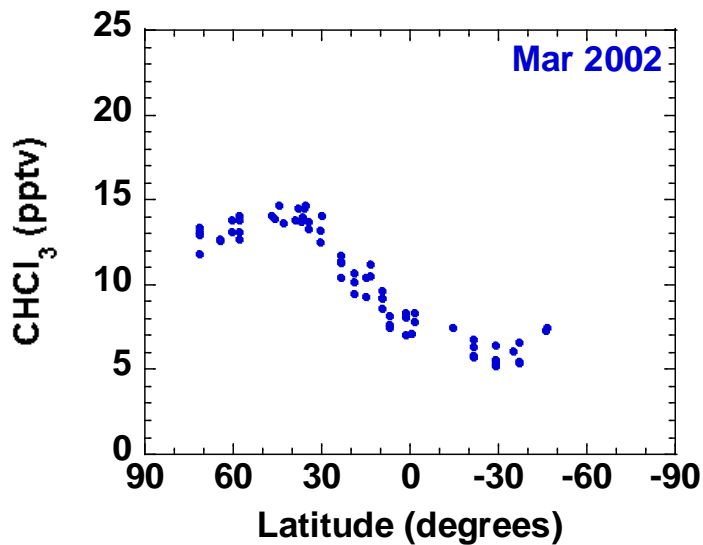
*Tetrachloroethene ( $C_2Cl_4$ ) is an industrial solvent (lifetime 3-4 mo). It is carcinogenic and affects the central nervous system.*



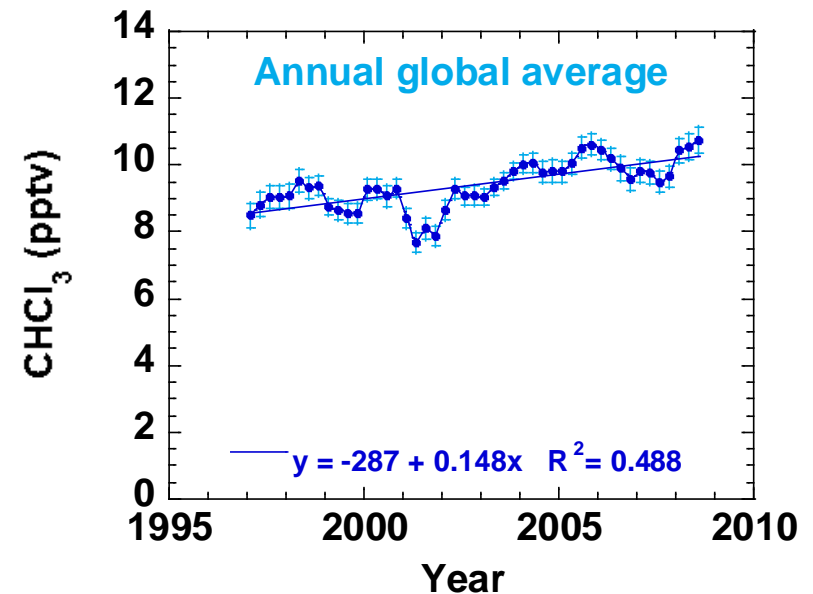
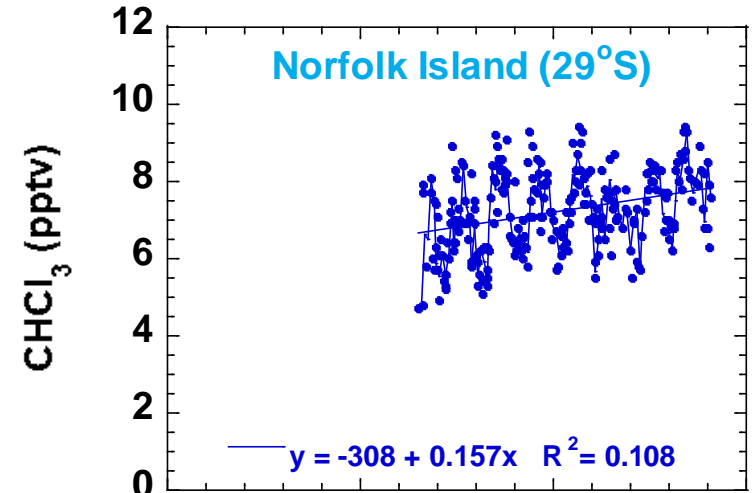
# Global $\text{CHCl}_3$ trend: Increasing

## Long-term global $\text{CHCl}_3$ increase

- $9.0 \pm 0.3$  pptv in 1997
- $10.7 \pm 0.4$  pptv in 2008
- ~ 20% increase in 11 years

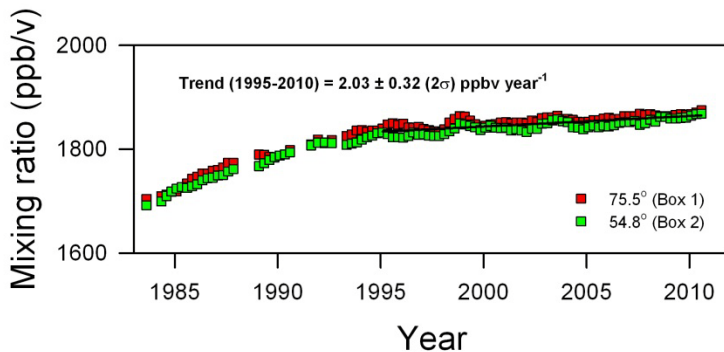
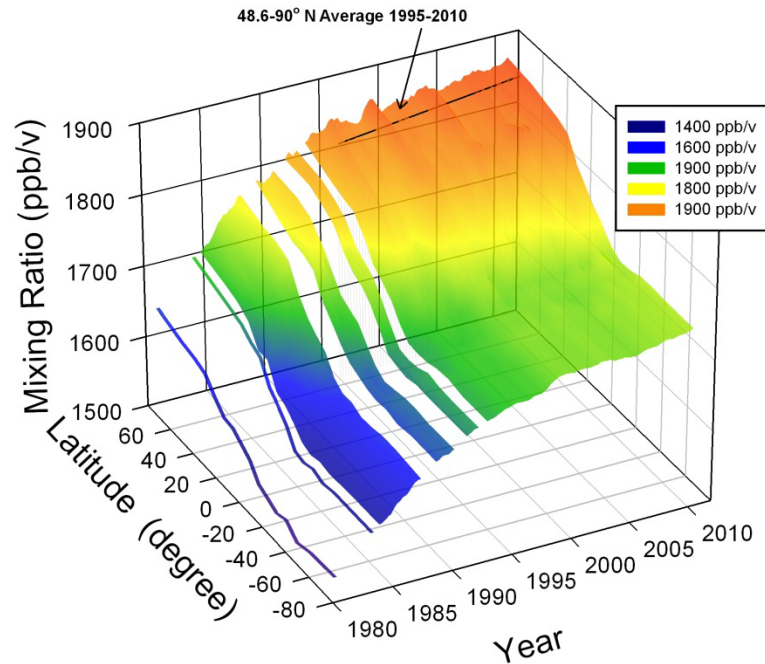


*Chloroform ( $\text{CHCl}_3$ ) is an industrial solvent (lifetime 3-5 mo). It is produced as a precursor to Teflon.*



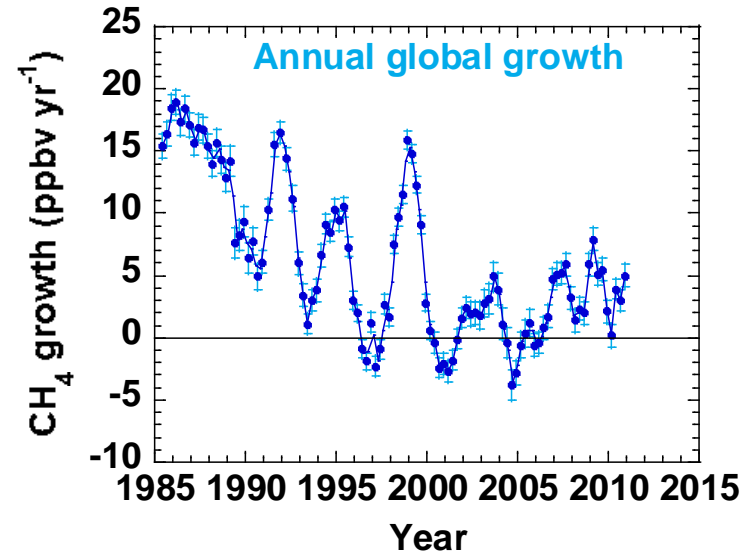
# Global CH<sub>4</sub> trend: Increasing, variable

## Annual CH<sub>4</sub> Trends



## Long-term global CH<sub>4</sub> increase

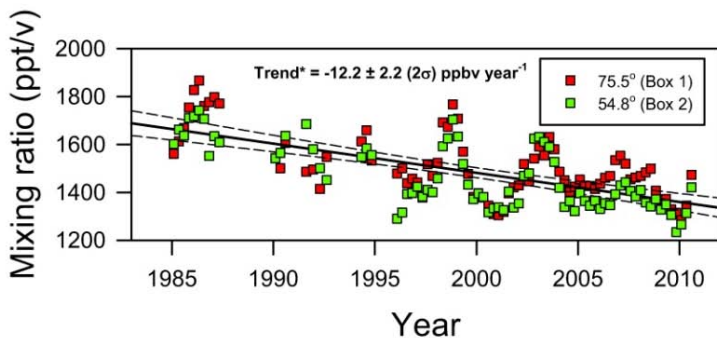
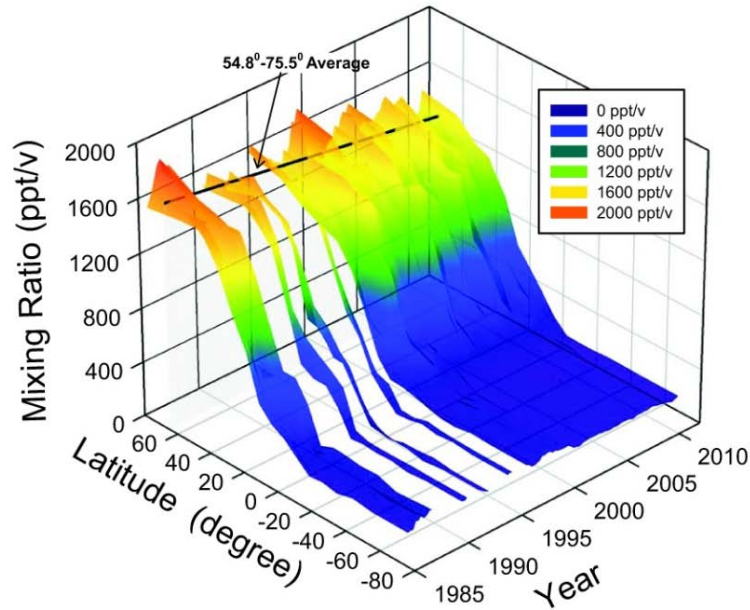
- $1568 \pm 2$  ppbv in 1980
- $1792 \pm 1$  ppbv in 2010
- **225 ppbv (14%) increase in 30 years**
- Growth in 2010:  $5.3 \pm 0.7$  ppbv yr<sup>-1</sup>



*Methane (CH<sub>4</sub>) is a potent greenhouse gas. Its sources include wetlands, rice paddies, fossil fuel and biomass burning (lifetime 9 years).*

# Global ethane trend: Declining

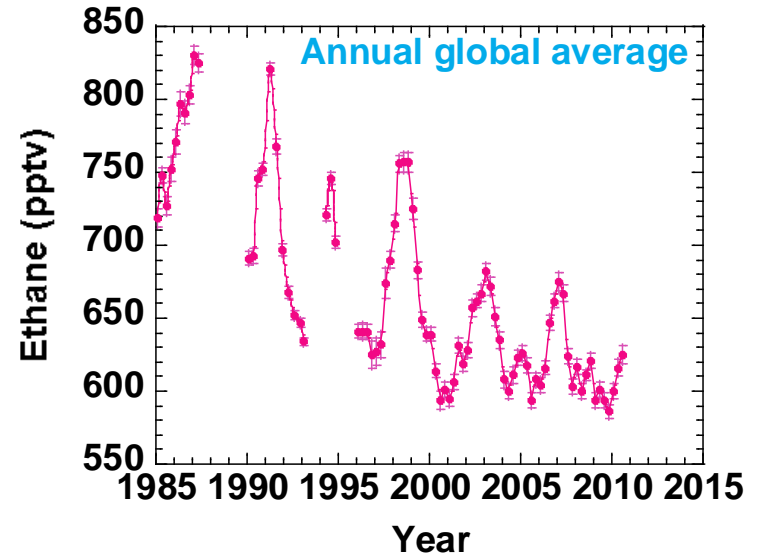
## Annual Ethane Trends



\* Dashed lines are 99% confidence intervals.

## Long-term global ethane decline

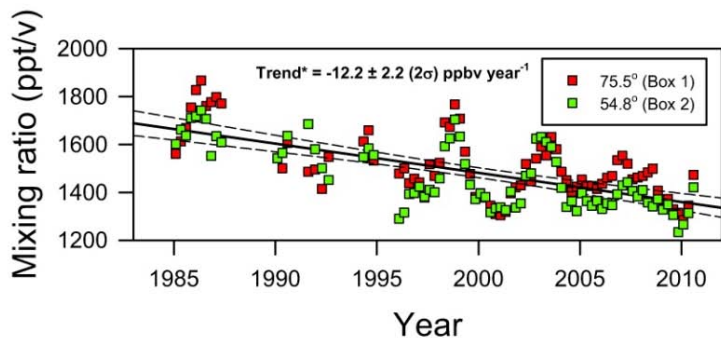
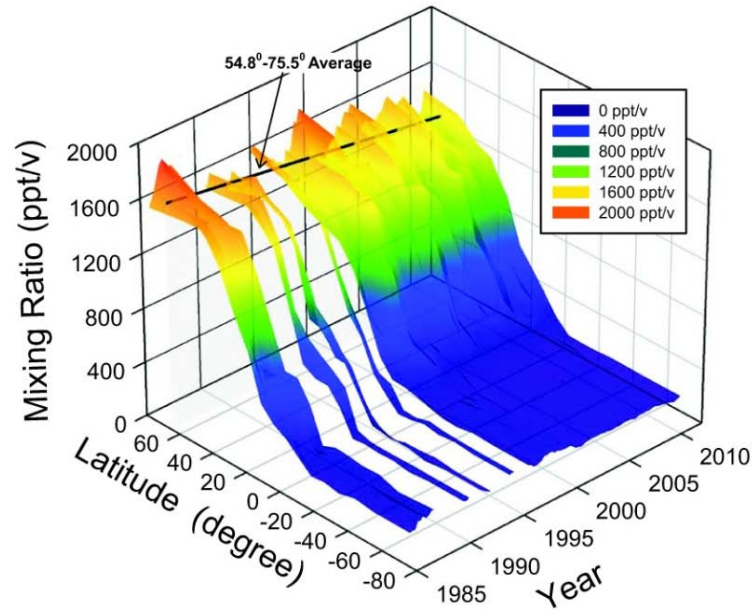
- $791 \pm 19$  pptv in 1986
- $625 \pm 10$  pptv in 2010
- **170 pptv (21%) decline in 25 years**
- Growth in 2010:  $31 \pm 11 \text{ pptv yr}^{-1}$



*Ethane ( $\text{C}_2\text{H}_6$ ) is emitted during fossil fuel production (unburned gas) and by biofuel and biomass burning (lifetime 2-3 months)*

# Global ethane trend: Declining

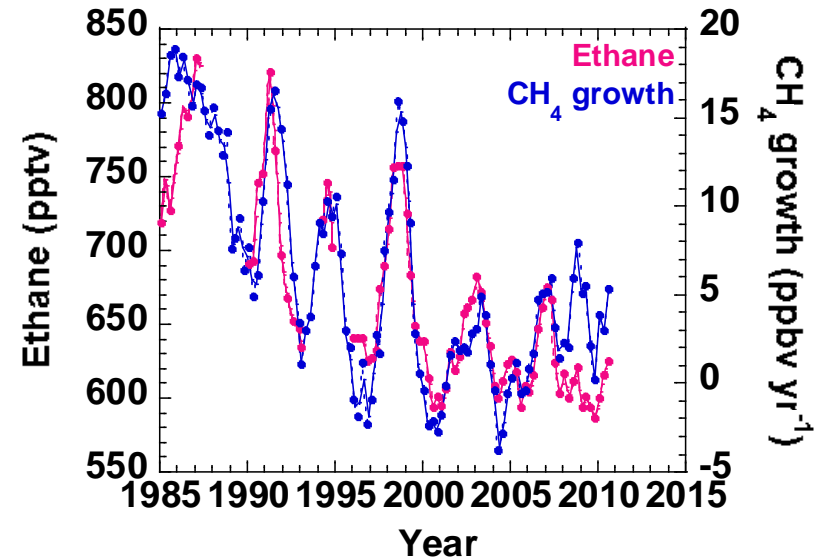
## Annual Ethane Trends



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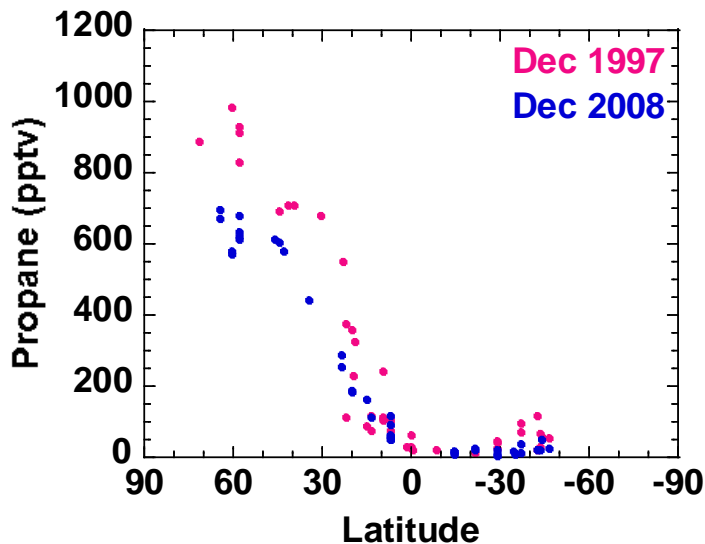


*Ethane ( $\text{C}_2\text{H}_6$ ) is emitted during fossil fuel production (unburned gas) and by biofuel and biomass burning (lifetime 2-3 months)*

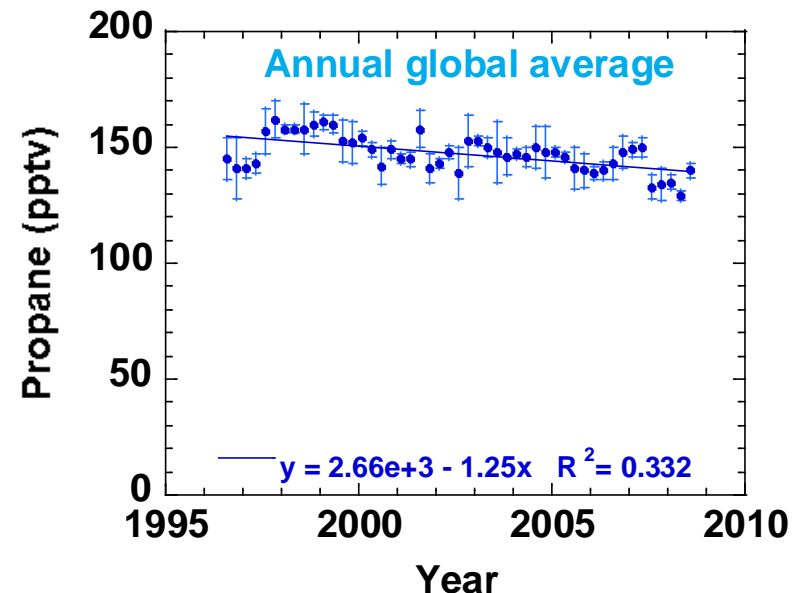
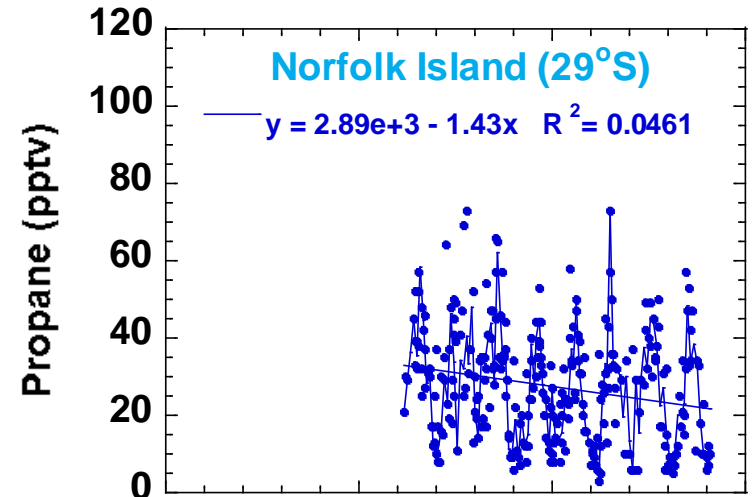
# Global propane trend: Decreasing

## Long-term global propane decrease

- $158 \pm 2$  pptv in 1998
- $140 \pm 2$  pptv in 2008
- ~ 10% decrease in 10 years



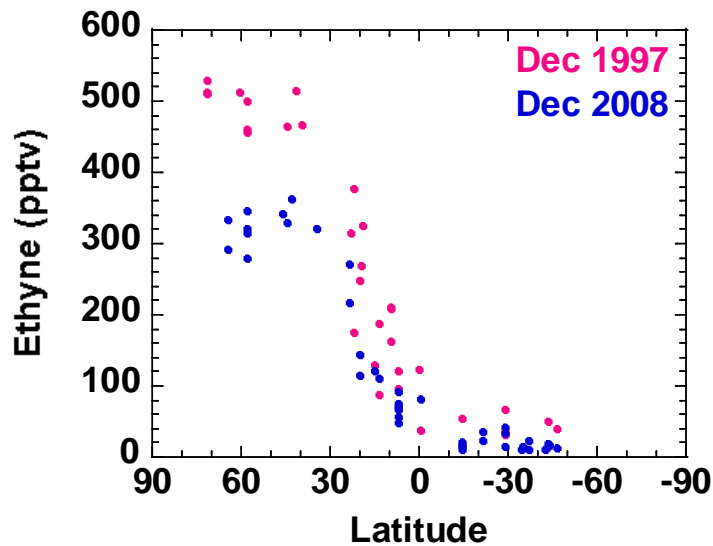
*Propane (C<sub>3</sub>H<sub>8</sub>) is emitted by fossil fuel production, biofuel and biomass burning, and gasoline exhaust (lifetime 1-2 weeks)*



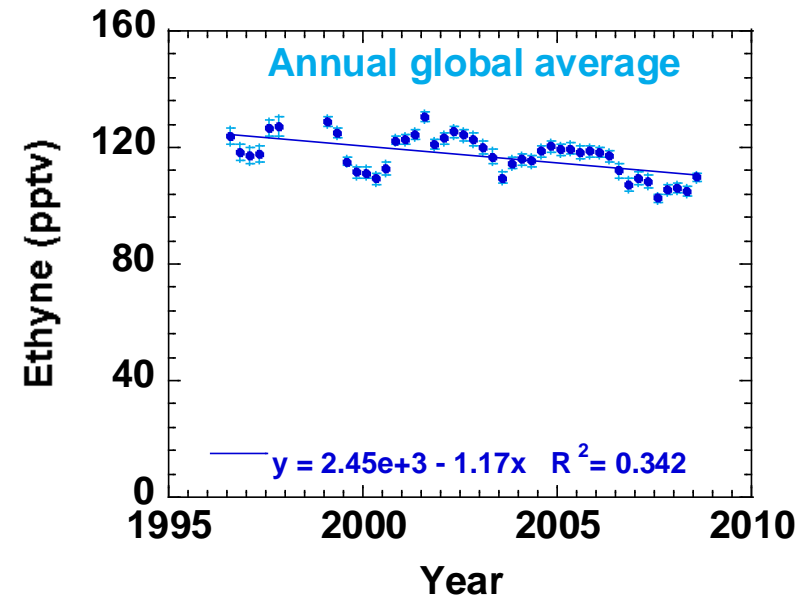
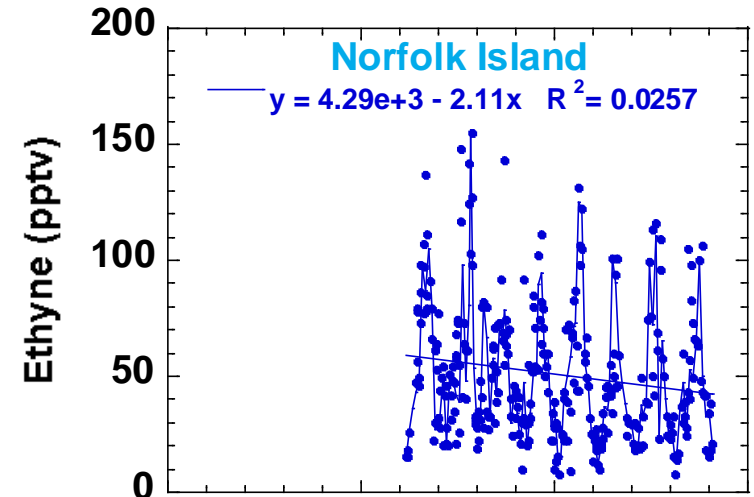
# Global ethyne trend: Decreasing

## Long-term global ethyne decrease

- $124 \pm 3$  pptv in 1996
- $110 \pm 2$  pptv in 2008
- **~ 11% decrease in 12 years**



*Ethyne ( $C_2H_2$ ) is a tracer of incomplete combustion by biomass burning and urban fossil fuel (lifetime 2-3 weeks).*



# Conclusions and Acknowledgments

## Multi-decade record shows significant trends in global trace gas mixing ratios:

- Global declines of many **halocarbons** in response to banning legislation
- Global declines of many **hydrocarbons** due to less venting/flaring; vehicle emission controls
- Global increases of **CHCl<sub>3</sub>**, **HCFCs** and **CH<sub>4</sub>**

### Global declines:

#### *Halocarbons*

- CFC-11
- CFC-12
- CFC-113
- H-1211
- CCl<sub>4</sub>
- CH<sub>3</sub>CCl<sub>3</sub>
- CFC-114
- C<sub>2</sub>Cl<sub>4</sub>

#### *Alkanes*

- Ethane
- *i*-Butane
- Propane
- *n*-Butane

#### *Alkynes*

- Ethyne

### Global increases:

#### *Halocarbons*

- HCFC-22
- HCFC-141b
- HCFC-142b
- CHCl<sub>3</sub>

#### *Alkanes*

- Methane

### *Acknowledgments*

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